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A new and versatile one-pot synthesis of indol-2-ones by a novel Ugi-four-component-Heck reaction

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Abstract—A novel one-pot-synthesis of highly substituted indol-2-ones using a combination of Ugi and Heck reaction (U-4CR-Heck) is described. The synthesized indol-2-ones represent an interesting pharmacological scaffold with four potential points of diversity. Thus, this novel reaction-type is amenable to combinatorial high-throughput screening. $© 2006 Elsevier Ltd. All rights reserved.$

Compounds containing the indol-2-one scaffold constitute an important pharmacophoric moiety, which exhibits important biological effects such as antitumor activity (1) ,^{[1,2](#page-1-0)} phosphodiesterase inhibitor activity (2) ,^{[3](#page-1-0)} and tyrosine kinase inhibitor activity (3) .^{[4–8](#page-1-0)} These examples illustrate the ongoing interest toward the scaffolds containing indol-2-ones (Fig. 1). However, the described synthetic strategies to indol-2-ones have found limitations mainly due to the lack of versatility and a limited number of appropriate initial reactants. In addition, they have not provided a robust method suitable for the production of combinatorial libraries.

Therefore we developed a novel one-pot synthesis of the indol-2-one core structure by combining combinatorial and classical sequential chemistry (Scheme 1). The presented novel Ugi–Heck reaction (U-4CR-Heck) describes an extension of former presented works.

Therein Gracias et al. 9 and Yang et al. 10 developed independently from each other an Ugi/Heck two-step

Scheme 1. One-pot indol-2-one synthesis via U-4CR-Heck.

Figure 1. Pharmacological interesting structures incorporating the indol-2-one motif.

Keywords: Ugi reaction; Heck reaction; One-pot synthesis; Indol-2-ones; Multicomponent reaction; Combinatorial chemistry. * Corresponding author. Tel.: +49 8158 90400; fax: +49 8158 904022; e-mail: umkehrer@priaton.de

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Table 1. Synthesized indol-2-ones

(a) Trifluorethanol; 24 h; 50 $^{\circ}$ C.

(b) Acetonitrile; 24 h; 80 °C; Pd(OAc)₂; PPh₃.

synthesis of N-heterocycles, using the advantages of multicomponent and classical sequential chemistry. The combination of multicomponent reactions and different post-reactions results in a maximum of diversity, which is well reported by different working $groups.^{11–17}$ $groups.^{11–17}$ $groups.^{11–17}$

Multicomponent reactions like the Ugi reaction allow rapid generation of compound libraries^{[18–24](#page-2-0)} and the final intramolecular Heck reaction is used for a defined ring-closing. The obtained high substituted indol-2-ones have four points of potential diversity. Thus this novel reaction type is amenable to high-throughput combinatorial library production.

The formation of the acyclic products (7) was originally reported by Ugi et al. and the final ring-closing was performed by a classical intramolecular Heck reac-tion^{[25–29](#page-2-0)} ([Scheme 1\)](#page-0-0). These two reaction steps were combined in a new one-pot synthesis. The Ugi–Heck reaction was performed in a typical procedure, 30 whereby the starting materials were mixed in equimolar amounts and stirred for 24 h at 50 °C. Afterwards solvent was changed from polar protic to polar aprotic and 10% of Pd-catalyst were added. The reaction mixture was stirred for another 24 h at 80 $^{\circ}$ C. The expected compounds (8a–m) were isolated as isomeric mixtures with moderate to good yields (Table 1). Their assignment was made on the base of 1 H NMR, 13 C NMR, and HPLC–MS data. All the synthesized compounds have purity >95%.

In this letter we show an efficient synthesis of various types of 1,3-disubstituted indol-2-ones (Table 1). The used aldehydes, anilines, isocyanides, and acrylic acids can be varied broadly, producing products with four potential points of diversity.

In summary, a novel one-pot solution phase procedure for the preparation of indol-2-ones has been reported. With final products containing four points of potential diversity and a facile and rapid production protocol, access to thousands of diverse analogues with the aforementioned core structure is now feasible.

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Current efforts are now focused on further extension of this novel reaction type.

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- 30. Typical procedure: The aldehyde (3 mmol) and the amine (3 mmol) were stirred in 5 ml trifluoroethanol for 1 h at room temperature. Then 3 mmol of acid and 3 mmol of isocyanide were added. The reaction mixture was stirred for 16 h at 50 °C until the reaction was completed (indication by TLC). After the reaction was completed, the solvent was evaporated and the resulting residue was dissolved in 5 ml of acetonitrile and 0.3 mmol Pd(OAc)₂

and 0.6 mmol PPh₃ were added. The reaction mixture was stirred for additional $16-24$ h at 80° C. Afterwards the solvent was evaporated again and the resulting residue was dissolved in ethyl acetate and filtered through a pad of silica. The resulting crude product was purified by crystallization from ethanol or by column chromatography on silica gel (hexane/ethyl acetate).

Compound 8c was isolated in 52% yield as a yellow solid (mp 156 °C). ¹H NMR (CDCl₃, 250.13 MHz): 3.69 (s, 3H, CH₃), 4.03 (d, ³J = 5.3 Hz, 2H, CH₂), 4.49 (s, 2H, CH₂), 6.87 (s, 1H, NH), 6.90–7.66 (m, 9H, C_6H_5), 7.84 (s, 1H, CH=C). ¹³C NMR (CDCl₃, 62.90 MHz): 41.1 (CH₂), 43.9 $(CH₂), 52.3 (CH₃), 109.0, 121.1 (Cq), 122.5, 122.8, 126.3)$ (Cq) , 128.6, 129.3, 130.0, 134.5 (Cq) , 138.3 $(CH=C)$, 142.5 (Cq), 167.5 (CON), 168.7 (CON), 169.8 (COOCH3). MS (ESI): $m/z = 351.1$ [M+H]⁺, 373.2 [M+Na]⁺.

Compound 8d was isolated in 54% yield as a yellow-brown solid (mp 194 °C). ¹H NMR (CDCl₃, 250.13 MHz): 1.34 $\{s, 9H, C(CH_3)_3\}, 6.27$ (s, 1H, CH–C₆H₅), 6.80–7.53 (m, 14H), 7.59 (s, 1H, CH=C), 8.23 (s, 1H, NH). ¹³C NMR $(CDCl_3, 62.90 MHz)$: 28.5 $\{C(CH_3)_3\}$, 51.9 $\{C(CH_3)_3\}$, 58.9 (CH–C₆H₅), 111.8, 118.8, 122.0, 124.2, 125.3, 128.1, 128.2, 128.6, 128.7, 128.8, 130.5, 131.9, 133.7, 134.6, 137.7, 140.4, 166.3 (CON), 167.1 (CON). MS (ESI): $m/z = 411.0$ $[M+H]^{+}$, 433.3 $[M+Na]^{+}$.

Compound 8g was isolated in 58% yield as a yellow solid (mp 251 °C). ¹H NMR (DMSO, 250.13 MHz): 4.32 $(d, {}^{3}J = 5.11 \text{ Hz}, 2H, CH_2), 4.49 \text{ (s, } 2H, CH_2), 6.89-7.48$ $(m, 9H)$, 7.79 (s, 1H, C=CH), 8.00 (d, $3J = 8.46$ Hz, 2H, $-C_6H_4-NO_2$), 8.37 (d, ³ $J = 8.46$ Hz, 2H, $-C_6H_4-NO_2$),
8.77 (t, 1H, ³ $J = 5.11$ Hz, $-CH_2-NH$). ¹³C NMR (DMSO, 62.90 MHz): 42.2 ($CH₂$), 42.5 ($CH₂$), 109.4, 119.8, 122.0, 123.2, 124.0, 126.8, 127.2, 128.3, 129.0, 130.5, 130.8, 132.5, 133.7, 139.1, 141.3, 144.1, 147.5, 165.2 (CON), 166.5 (CON), 167.0 (CON). MS (ESI): $m/z =$ 414.5 $[M+H]$ ⁺, 436.5 $[M+Na]$ ⁺.

Compound $8l$ was isolated in 45% yield as a yellow oil. ¹H NMR (CDCl₃, 250.13 MHz): 1.94 (d, ³J = 7.0 Hz, 3H, CH₃-CH), 3.78 (s, 3H, COOCH₃), 4.10 (d, ³J = 5.24 Hz, 2H, CH₂), 4.67 (s, 2H, CH₂), 6.69 (s, 1H, NH), 6.87–7.43 (m, 5H). ¹³C NMR (CDCl₃, 62.90 MHz): 18.2 (CH₃), 40.8 $(CH₂), 43.6 (CH₂), 52.5 (COOCH₃), 108.6, 119.1, 121.3,$ 122.7, 127.0, 128.8, 138.9, 147.2, 167.5 (CON), 167.7 (CON), 170.0 (COOCH₃). MS (ESI): $m/z = 289.1$
[M+H]⁺, 311.3 [M+Na]⁺.